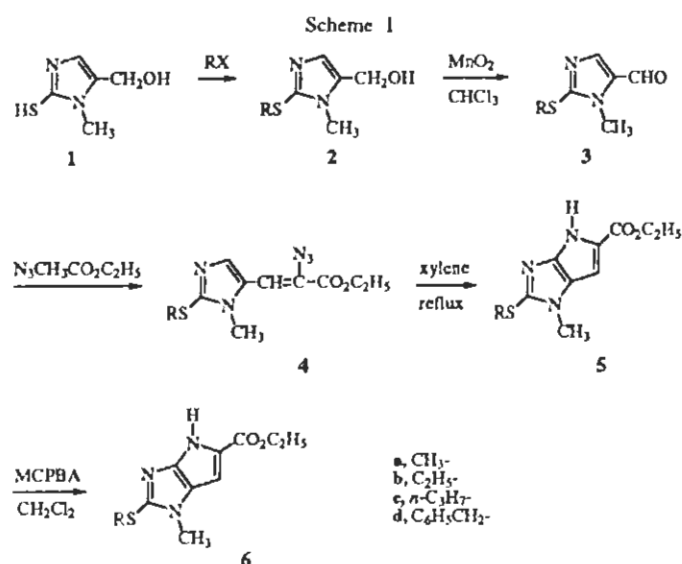


Department of Chemistry, Faculty of Pharmacy, The Medical Sciences University
of Tehran, Tehran, Iran
Received September 6, 1996

Starting from the readily available 5-hydroxymethyl-2-mercapto-1-methylimidazole (1) substituted pyrrolo[2,3-*d*]imidazoles were prepared.

J. Heterocyclic Chem., 34, 549 (1997).

In view of the potential biological activity of fused imidazoles [1,2] the syntheses of the title compounds as possible effective drugs against tropical diseases [3] were of interest to us. The title compounds were prepared according to Scheme 1.



Reaction of 5-hydroxymethyl-2-mercapto-1-methylimidazole (1) [4] with alkyl halides afforded corresponding substituted alkylthioimidazoles 2 [5,6]. Oxidation of compounds 2 with manganese dioxide in chloroform gave 2-alkylthio-5-formyl-1-methylimidazoles 3 [6]. Condensation of compound 3 with ethyl azidoacetate according to the procedure reported previously [7], afforded ethyl α -azido- β -(2-alkylthio-1-methylimidazol-5-yl)acrylates 4. Cyclization of compounds 4 to ethyl 2-alkylthio-1-methylpyrrolo[2,3-*d*]imidazole-5-carboxylates 5 was accomplished through boiling in xylene [7,8]. Compounds 5 was oxidized with *m*-chloroperoxybenzoic acid to the desired final compounds, namely ethyl 2-alkylthio-1-methylpyrrolo[2,3-*d*]imidazole-5-carboxylates 6 [5].

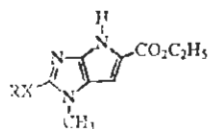
The physical constants of compounds 5 and 6 are summarized in Table 1.

The structure of all compounds were confirmed by elemental analysis, ir, nmr and mass spectroscopy.

EXPERIMENTAL

Melting points were determined on a Kofler hot stage apparatus and are uncorrected. The ir spectra were obtained using a Perkin-Elmer Model 267 spectrograph. The ¹H-nmr spectra were recorded on a Bruker FT-80 spectrometer and chemical shifts (δ) are in ppm relative to internal tetramethylsilane. The

Table 1



Compound	Yield No.	R (%)	X	T _m , °C [a]	Formula	C%		H%		N%	
						Calcd.	Found	Calcd.	Found	Calcd.	Found
5a	28	CH ₃	S	144-145	C ₁₀ H ₁₃ N ₃ O ₂ S	50.21	50.36	5.44	5.59	17.57	17.48
5b	27	C ₂ H ₅	S	80-82	C ₁₁ H ₁₅ N ₃ O ₂ S	52.17	52.02	5.93	6.05	16.60	16.49
5c	20	C ₃ H ₇	S	85-87	C ₁₂ H ₁₇ N ₃ O ₂ S	53.93	53.99	6.37	6.31	15.73	15.67
5d	38	C ₆ H ₅ CH ₂	S	155-158	C ₁₆ H ₁₇ N ₃ O ₂ S	60.95	60.91	5.40	5.22	13.33	13.19
6a	62	CH ₃	SO ₂	181-182	C ₁₀ H ₁₃ N ₃ O ₄ S	44.28	44.37	4.80	4.61	15.50	15.35
6b	72	C ₂ H ₅	SO ₂	164-165	C ₁₁ H ₁₅ N ₃ O ₄ S	46.32	46.49	5.26	5.40	14.74	14.63
6c	61	C ₃ H ₇	SO ₂	167-168	C ₁₃ H ₁₇ N ₃ O ₄ S	50.16	50.35	5.47	5.75	13.50	13.61
6d	62	C ₆ H ₅ CH ₂	SO ₂	180-183	C ₁₆ H ₁₇ N ₃ O ₄ S	55.33	55.43	4.90	5.10	12.10	12.25

[a] All compounds were crystallized from ether.

mass spectra were run on a Varian Model MAT-MS 311 spectrometer at 70 eV.

2-Ethylthio-1-methyl-5-hydroxymethylimidazole (2b).

To a stirring suspension of compound 1 (5 g, 34.72 mmol) in methanol (500 ml) is added dropwise 36 ml of 1.0 N sodium hydroxide at room temperature. The clear pale yellow suspension is stirred for 10 minutes. Iodoethane (3 ml, 36.75 mmol) is added dropwise and stirring is continued overnight. After evaporation of the methanol the residue was dissolved in water and extracted with chloroform. The solvent was evaporated and the residue was crystallized from ether to give 4 g (80%) of compound 2b, mp 48–50°; ir (potassium bromide): ν 3291 (OH); ^1H nmr (deuteriochloroform): 6.61 (s, 1H, H-C₄ imidazole), 4.59 (s, 2H, CH₂OH), 3.64 (s, 3H, NCH₃), 3.04 (q, 2H, J = 7.2 Hz, CH₂S), and 1.33 ppm (t, 3H, J = 7.20 Hz, CH₃); ms: m/z (%) 172 (M⁺, 84), 156 (16), 143 (34), 139 (100), 97 (10), 74 (15), 42 (24).

Anal. Calcd. for C₇H₁₂N₂OS: C, 48.84; H, 6.98; N, 16.27. Found: C, 48.88; H, 6.80; N, 16.13.

Other 2-alkylthio-5-hydroxymethyl-1-methylimidazoles were prepared similarly.

2-Ethylthio-5-formyl-1-methylimidazole (3b).

A stirring suspension of compound 1 (3.56 g, 20.7 mmol) and manganese dioxide (11.6 g, 133.2 mmol) in chloroform (70 ml) was refluxed for 12 hours. The reaction mixture was cooled to room temperature and filtered. The chloroform was evaporated to give 3.27 g (93%) of an oil; ir: ν 1663 cm⁻¹ (C=O); ^1H nmr: 9.58 (s, 1H, CH=O), 7.73 (s, 1H, H-C₄ imidazole), 3.82 (s, 3H, NCH₃), 3.29 (q, 2H, CH₂S) and 1.42 ppm (t, 3H, CH₃); ms: m/z (%) 171 (M⁺+1, 100), 170 (M⁺, 59), 137 (22), 114 (18).

Other 2-alkylthio-5-formyl-1-methylimidazoles were prepared similarly.

Ethyl α -Azido- β -(2-ethylthio-1-methylimidazol-5-yl)acrylate (4b).

To a stirring solution of sodium (1.38 g, 59.75 mmol) in absolute ethanol (37.5 ml) at -20° was added a solution of compound 3b (2.5 g, 19.75 mmol) and ethyl azidoacetate (7.63 g, 58.88 mmol) in dry THF (30 ml) and absolute ethanol (30 ml). After 2 hours at -10°, the mixture was added to a saturated solution of ammonium chloride. The mixture was extracted with ether. The ether was evaporated and the residue was purified with column chromatography (silica gel, petroleum ether as eluent) to give 3 g (72.5%) of an oily compound 4b; ir: ν 2119 (azide), 1712 cm⁻¹ (C=O); ^1H nmr (deuteriochloroform): 7.91 (s, 1H, H-C=C), 6.67 (s, 1H, H-C₄ imidazole), 4.36 (q, 2H, CO₂CH₂, J = 7.09 Hz), 3.55 (s, 3H, NCH₃), 3.17 (q, 2H, CH₂S, J = 7.31 Hz) and 1.37 ppm (t, 3H, CH₃); ms: m/z (%) 282 (M⁺+1, 6), 224 (12), 180 (100), 152 (41), 111 (8), 81 (2).

Other ethyl α -azido- β -(2-alkylthio-1-methylimidazol-5-yl)acrylates were prepared similarly.

Ethyl 2-Ethylthio-1-methylpyrrolo[2,3-d]imidazole-5-carboxylate (5b).

A solution of compound 4b (0.4 g, 1.42 mmol) in xylene (20 ml) was refluxed for 2 hours. The solvent was evaporated and the residue was purified with column chromatography (silica gel, chloroform as eluent) to give 0.1 g (27%) of compound 5b, mp 80–82° (ether); ir (potassium bromide): ν 3460 (NH), 1700 cm⁻¹ (C=O); ^1H

nmr (deuteriochloroform): 9.54 (bs, 1H, NH), 6.67 (s, 1H, H-C₆), 4.35 (q, 2H, CO₂CH₂), 3.67 (s, 3H, NCH₃), 3.22 (q, 2H, CH₂S) and 1.36 ppm (m, 6H, CH₃); ms: m/z (%) 253 (M⁺, 100), 224 (43), 220 (17), 178 (71), 174 (21), 150 (9), 109 (17).

Anal. Calcd. for C₁₁H₁₅N₃O₂S: C, 52.17; H, 5.93; N, 16.60. Found: C, 52.02; H, 6.05; N, 16.49.

Other ethyl 2-alkylthio-1-methylpyrrolo[2,3-d]imidazole-5-carboxylates were prepared similarly (Table 1).

Ethyl 2-Ethylsulfonyl-1-methylpyrrolo[2,3-d]imidazole-5-carboxylate (6b).

To a stirring solution of compound 5b (0.47 g, 1.086 mmol) in methylene chloride (30 ml) at 0° was added sodium bicarbonate (1.06 g, 9.43 mmol), followed by *m*-chloroperbenzoic acid (1.06 g, 4.92 mmol). The reaction mixture was stirred at 0° for 2 hours, then at room temperature overnight. An additional amount of *m*-chloroperbenzoic acid (0.1 g, 0.57 mmol) is added, and stirring is continued for 6 hours. The mixture is poured into water. The organic layer was washed with aqueous sodium bicarbonate followed by water.

The organic layer was evaporated and the residue was purified by column chromatography (silica gel, chloroform as eluent) to give 0.38 g (72%) of compound 6b, mp 164–165° (ether); ir (potassium bromide): ν 3340 (NH), 1700 cm⁻¹ (C=O); ^1H nmr (deuteriochloroform): 9.33 (bs, 1H, NH), 6.77 (s, 1H, H-C₆), 4.39 (q, 2H, CO₂CH₂), 4.09 (s, 3H, NCH₃), 3.53 (q, 3H, CH₂S) and 1.40 ppm (m, 6H, CH₃); ms: m/z (%) 285 (M⁺, 100), 240 (21), 208 (62), 193 (96), 118 (20), 91 (36), 66 (14).

Anal. Calcd. for C₁₁H₁₅N₃O₄S: C, 46.32; H, 5.26; N, 14.74. Found: C, 46.49; H, 5.40; N, 14.63.

Other ethyl 2-alkylsulfonyl-1-methylpyrrolo[2,3-d]imidazole-5-carboxylates were prepared similarly (Table 1).

Acknowledgement.

This work was partially supported by a grant from the research council of the Medical Sciences University of Tehran and the International Organization for Chemical Sciences in Development (IOCD).

REFERENCES AND NOTES

- [1] M. Fortin, D. Frechet, G. Hamon, S. Jouquey and J. P. Vever, *European Patent Appl.* 464,040 (1991); *Chem. Abstr.*, **116**, 151760m (1992).
- [2] E. Guzman, M. Cain, P. Larscheid, T. Hagen, J. M. Cook, N. Schwen, P. Skolnick and S. M. Paul, *J. Med. Chem.*, **27**, 563 (1984).
- [3] G. T. Seaborg, *Science*, **223**, 9 (1984).
- [4] J. M. Dener, L. H. Zhang, and H. Rapoport, *J. Org. Chem.*, **58**, 1156 (1993).
- [5] J. J. Paddon, P. K. Lumma, G. S. Ponticello, E. G. Novello and J. N. Sprague, *J. Heterocyclic Chem.*, **14**, 889 (1977).
- [6] N. J. P. Broom, P. J. O'Tuallan, and J. S. Elder, *PCT Int. Appl.* WO 93 06,118 (1993); *Chem. Abstr.*, **119**, 203236e (1993).
- [7] A. Shafiee, A. Mazioumi and V. I. Cohen, *J. Heterocyclic Chem.*, **16**, 1563 (1979).
- [8] E. V. Sadanandan, S. K. Pillai, M. V. Lakshminathan, A. D. Billimoria, J. S. Culpepper, M. P. Cava, *J. Org. Chem.*, **60**, 1800 (1995).